Optimization of flotation parameters for talc carbonates of Mingora emerald mine (Swat), Khyber Pakhtunkhwa, Pakistan

Nisar Mohammad, Sajjad Hussain and Noor Mohammad

Department of Mining Engineering, UET Peshawar, Khyber Pakhtunkhwa, Pakistan

Abstract

The aim of the research work is to optimize parameters affecting efficiency of flotation process for obtaining high quality talc from talc carbonates of Mingora emerald mines, Swat, KP. Different parameters of flotation process like, pH, pulp density, depressant dosage, collector dosage, were optimized for maximum grade and recovery of talc. The collector used was a mixture of equal amounts of oleic acid and kerosene oil, Depressant used was sodium hexa metaphosphate and frother used was propyl glycol. Dosage of the collector was varied from 0.6 to 1.4 kg/ton, Depressant dosage varied from 0.4 to 1.2 kg/ton and pH of the pulp was varied as 4, 7, 9, 11 and 12. Optimum grade and recovery of talc was achieved at pH 11, depressant dosage 1.0 kg/ton, collector dosage as 1.2 kg/ton, pulp density 200 gram/liter (g/l), at constant impeller speed of 1100 rpm and frother dosage of 0.1 kg/ton.

Keywords: Talc; Flotation; Recovery; Grade; Depressant; Emerald mine.

1. Introduction

The Swat emerald deposits are situated in the close proximity of Mingora town in district Swat. These deposits are at a distance of about 130 miles to the northeast of Peshawar and 200 miles northwest of Islamabad. Swat emerald mine is approachable by a good metaled road. The Mingora emerald deposits are commonly known as the Swat emerald mines (Muhammad, 2011). The host rock of emerald crystal in the deposit is talc carbonate schist. After excavation for emerald, the talc bearing host rock is thrown away as a refuse because of its low grade. Being a useful and demanded commodity the authors felt the need to take research on processing of this by- product (raw talc) of emerald mines to make it industrial grade product through flotation process so as to save the foreign exchange being spent on import of talc and talc products.

Talc is one of the naturally floatable mineral. It is mostly associated with gangue materials which determine its application as industrial mineral. Both chemical composition and physical properties like whiteness, bulk density, grain size etc. of talc are important. Talc is hydrous magnesium silicate with chemical formula Mg₃ Si₄ O₁₀ (OH)₂. When pure, it contains 63.36% silica, 31.89% magnesia and 4.75 % water of crystallization

(Siddiqui, 1993). The major elemental impurities in talc understudy were CaO, Al_2O_3 , and Fe_2O_3 . Various processing techniques are in use for removal of the associated gangue minerals with talc like selective mining, hand/optical sorting, differential grinding and sizing, magnetic separation, gravity separation, pneumatic separation, flotation and leaching. Processing studies have been conducted in different parts of the world for evaluation and up gradation of talc ores with varying mineralogical composition and with wide range of beneficiation techniques being adopted.

As talc is naturally floatable, therefore, worldwide most of the researchers have used flotation and leaching methods and combination of both methods for talc ore having carbonate impurities at different leaching and flotation parameters (Ge Li, 2013; Hojamberdiev et al., 2010; Ahmed et al., 2007; Kursun and Ulusoy, 2006; Akdemir, 2005; Derco and Nemeth, 2002) . So in the light of literature review flotation process was used in present research work for talc separation from the associated impurities mostly carbonates.

2. Materials and methods

Five (5) rich talc samples in lumps form were selectively collected from different excavated talc carbonate dumps, weighing about 40 Kg as a whole and about 8 Kg of each sample from Mingora emerald mine, District Swat, KP. The five samples were crushed, ground, blended and prepared in the size range of -45µm. Single step bench scale flotation tests were conducted on the blended talc carbonate samples with varying flotation parameters like pH of pulp, pulp density, effect of collector and depressant dosage were studied to obtain optimum grade and recovery of talc.

3. Experimental work

3.1. Results and discussions

Chemical analysis results of feed/ head talc carbonate samples collected from Mingora town in district Swat presented in Table 2.

Contents	%
SiO ₂	44.74
MgO	37.22
CaO	0.92
Al ₂ O ₃	1.42
Fe ₂ O ₃	4.50

Table 1. Chemical analysis of head sample.

3.2. Flotation parameters tested

Mass and component recoveries of the flotation test products were calculated using the mass percent and assays of different constituents in the feed, concentrate and tailing by using the following formulas (Wills, 1992):

Mass recovery of the concentrate (Rm (c)) = (Concentrate mass /Feed mass)*100

Mass recovery of tailings (Rm (t)) = (Tailing mass /Feed mass)*100

Component recovery in concentrate (Rc (c)) = (Component mass in concentrate/ Component mass in feed)*100

Component recovery in tailings (Rc (t)) = (Component mass in tailing/ Component mass in feed)*100

3.2.1. Optimization of the pH value in talc flotation test

Five flotation tests were carried out at

different pH values, and at constant dosage of collector (1.0 kg/t), depressant (0.8 kg/t), and pulp density (200 g/l). The Chemical analysis results of flotation tests product are presented in Table 2a, which revealed that assay value of SiO₂ and MgO decreases with increase in pH value from 4 to 12, and percentage of elemental impurities increases with increasing pH value. The component recovery and mass recovery is given in Table 2b, which shows that, when the value of pH increases from 4 to 11, the component recovery of SiO₂ and MgO increased, the component recovery of elemental impurities increases and the mass recovery also increases. While the value of pH increased from 11and 12, the component recovery of SiO₂, MgO decreasesd, the component recovery of elemental impurities increases and mass recovery decreases. The optimum grade and recovery of talc are obtained at pH value 11 which assured that 11 is the optimum pH value.

The component and mass recoveries at different pH values are presented in Figure 1.

3.2.2. Optimization of the depressant dosage effect in talc flotation test

Chemical analysis of products that obtained through five flotation tests conducted at constant dosage of collector (1.0 kg/t), pulp density (200 g/l), Ph value (11) and at different dosage of depressant (hexa metaphosphate) are presented in Table 3a. The table shows that the assay value of SiO₂ and MgO decreased with increases the dosage of depressant from 0.4 to 1.2 kg/t, and the percentages of elemental impurities increased with increasing in dosage of depressant. The component and mass recoveries are given in Table 3b, which shows that the component recovery of SiO₂, MgO and elemental impurities increases, and mass recovery also increases, when the dosage of depressant increased from 0.4 to 1kg/t. The component recovery of SiO₂ and MgO and mass recovery of concentrate started decreasing when the dosage of depressant increased from 1.0 kg/t, which increased the component recovery of elemental impurities. This means that 1.0 kg/t is the optimum dosage of depression for obtaining the optimum talc recovery and grade.

Exp.	pН	Concer	ntrate				Tailings						
No.		Assay ((Ac) &				Assay (At), %						
		SiO ₂	MgO	CaO	Al_2O_3	Fe ₂ O ₃	SiO ₂	MgO	CaO	Al ₂ O ₃	Fe ₂ O ₃		
1	4	56.31	36.17	0.39	1.04	2.98	47.63	30.24	1.35	5.57	5.88		
2	7	55.74	35.91	0.47	1.17	3.13	46.54	26.91	1.44	6.24	6.89		
3	9	54.67	35.44	0.51	1.23	3.29	47.28	27.54	1.62	6.93	6.08		
4	11	53.59	34.82	0.58	1.31	3.57	41.91	20.48	1.94	9.74	7.63		
5	12	52.91	34.11	0.69	1.39	4.28	50.34	25.73	1.19	5.89	4.94		

Table 2a. Assay values of flotation tests products at different pH values.

Table 2b. Component and mass recoveries of flotation tests products at different pH values.

pН	Concer	ıtrate					Tailing	s				
	Compo	nent recov	very Rc(c)	,%		Rm(c),	Component recovery Rc(t), %					Rm(t),
	SiO ₂	MgO	CaO	Al ₂ O ₃	Fe ₂ O ₃	%	SiO ₂	MgO	CaO	Al ₂ O ₃	Fe ₂ O ₃	%
4	65.70	50.73	22.13	38.23	34.57	52.2	34.30	49.27	77.87	61.77	65.43	47.8
7	76.99	59.62	31.57	50.92	42.99	61.8	23.01	40.38	68.43	49.08	57.01	38.2
9	79.79	62.18	36.20	56.56	47.74	65.3	20.21	37.82	63.80	43.44	52.26	34.7
11	98.22	76.71	51.70	75.65	65.05	82.0	1.78	23.29	48.30	24.35	34.95	18.0
12	87.04	67.45	55.20	72.05	70.00	73.6	12.96	32.55	44.80	27.95	30.00	26.4

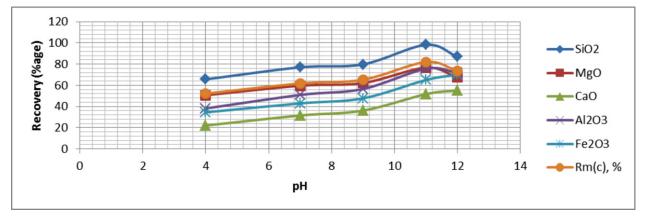


Fig. 1. Component and mass recoveries at different pH values.

Exp. No.	Depressant	Concer	ntrate			Tailings					
INU.	dosage, kg/t	Assay ((Ac) &		Assay (At), %						
		SiO ₂	MgO	CaO	Al_2O_3	Fe ₂ O ₃	SiO ₂	MgO	CaO	Al_2O_3	Fe ₂ O ₃
1	0.4	57.14	39.86	0.86	1.08	2.89	46.74	21.79	1.16	13.46	12.24
2	0.6	56.21	39.11	0.77	1.19	2.97	42.38	17.92	1.51	15.69	12.87
3	0.8	54.93	38.61	0.69	1.29	3.29	41.96	14.18	2.48	16.84	11.96
4	1.0	54.23	37.83	0.54	1.33	3.36	37.00	15.76	3.34	19.98	13.02
5	1.2	53.96	37.14	0.76	1.38	4.43	50.68	26.48	1.14	6.47	7.44

Table 3a Assay values of flotation tests	products at different depressant dosage
Table 3a. Assay values of flotation tests	products at unificient depressant dosage.

Depressant	Concer	ntrate					Tailings					
dosage, kg/t	Compo	nent recov	ery Rc(c)	,%		Rm(c),	Compo	Rm(t),				
0.4	SiO ₂	MgO	CaO	Al ₂ O ₃	Fe ₂ O ₃	%	SiO_2	MgO	CaO	Al ₂ O ₃	Fe ₂ O ₃	%
0.4	78.16	65.54	57.21	46.55	39.30	61.2	21.84	34.46	42.79	53.45	60.70	38.8
0.6	83.49	69.82	55.62	55.69	43.86	66.45	16.51	30.18	44.38	44.31	56.14	33.5
0.8	86.36	72.97	52.76	63.90	51.43	70.34	13.64	27.03	47.25	36.10	48.57	29.6
1.0	97.65	81.88	47.29	75.45	60.15	80.56	2.35	18.12	52.71	24.55	39.85	19.4
1.2	90.18	74.61	61.77	72.66	73.61	74.77	9.82	25.39	38.23	27.34	26.39	25.23

Table 3b. Component and mass recoveries of flotation tests products at different depressant dosage.

The component and mass recoveries at different depressant dosage given in Figure 2.

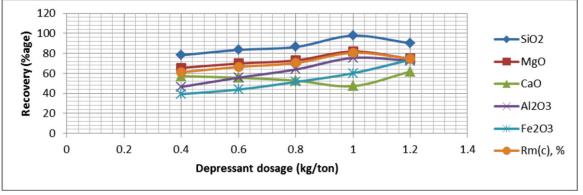


Fig. 2. Component and mass recoveries at different depressant dosage	Fig.	2.	Component a	and mas	s recoveries	at different	depressant	dosage.
--	------	----	-------------	---------	--------------	--------------	------------	---------

Table 4a. Assay values of f	lotation tests products at	different collector dosage.
-----------------------------	----------------------------	-----------------------------

Exp. No.	Collector	Concer	ntrate				Tailings				
INU.	dosage, kg/t	Assay ((Ac) &			Assay (At), %					
		SiO ₂	MgO	CaO	Al ₂ O ₃	Fe ₂ O ₃	SiO_2	MgO	CaO	Al ₂ O ₃	Fe ₂ O ₃
1	0.6	56.08	39.47	0.36	1.19	2.88	41.38	20.43	2.27	10.17	11.64
2	0.8	54.79	38.89	0.44	1.25	3.17	44.66	19.37	2.66	13.92	10.23
3	1.0	53.87	38.24	0.55	1.31	3.24	40.58	17.06	3.76	21.63	12.37
4	1.2	53.64	37.63	0.57	1.36	3.91	39.86	10.11	5.77	22.00	1591
5	1.4	52.91	36.86	0.62	1.40	4.19	46.28	26.74	2.41	10.24	7.73

Table 4b. Component and mass recoveries of flotation tests products at different col	ollector dosage.
--	------------------

Collect	Concen	itrate					Tailings					
or dosage,	Compo	nent reco	very Rc(:), %		Rm(c),	Component recovery Rc(t), %					Rm(t),
kg/t	SiO ₂	Mg	Ca	Al ₂ O	Fe ₂ O	%	SiO	Mg	Ca	Al ₂ O	Fe ₂ O	%
		0	0	3	3		2	0	0	3	3	
0.6	83.29	70.47	26.00	55.69	42.53	66.45	16.71	29.53	74.00	44.31	57.47	33.55
0.8	83.69	71.41	32.68	60.16	48.14	68.34	16.31	28.59	67.32	39.84	51.86	31.66
1.0	91.11	77.74	45.24	69.81	54.48	75.67	8.89	22.26	54.76	30.19	45.52	24.33
1.2	98.58	83.13	50.94	78.75	71.44	82.22	1.42	16.87	49.06	21.25	28.56	17.78
1.4	92.63	77.57	52.79	77.23	72.93	78.33	7.37	22.43	47.21	22.77	27.07	21.67

The component and mass recoveries at different collector dosage given in Figure 3.

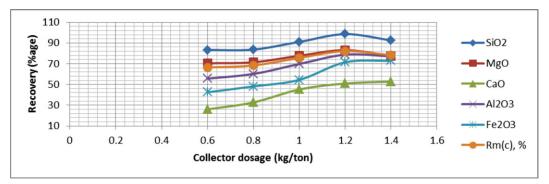


Fig. 3. Component and mass recoveries at different collector dosage.

Exp.	Pulp density,	Concer	ntrate			Tailings						
No.	g/L	Assay ((Ac) &				Assay (At), %					
		SiO ₂	MgO	CaO	Al_2O_3	Fe ₂ O ₃	SiO ₂	MgO	CaO	Al_2O_3	Fe ₂ O ₃	
1	100	56.43	35.62	0.33	0.96	2.90	42.36	24.31	2.37	11.35	12.26	
2	150	55.71	36.64	0.38	1.04	2.94	41.44	20.54	3.14	15.84	13.86	
3	200	54.86	36.95	0.48	1.13	3.04	38.67	14.66	5.66	22.63	15.47	
4	250	54.23	37.34	0.59	1.27	3.89	49.35	24.74	2.16	12.93	7.11	
5	300	53.07	30.24	0.75	1.39	4.19	53.71	26.94	1.73	7.04	5.24	

Table 5a. Assay values of flotation tests products at different pulp densities.

Table 5b. Component and	mass recoveries of flota	tion tests products at	different pulp density.
		reason reasons in	

Pulp density, g/L	Concentrate					Tailings						
	Component recovery Rc(c), %					Rm(c), %	Component recovery Rc(t), %				Rm(t), %	
	SiO ₂	MgO	CaO	Al_2O_3	Fe ₂ O ₃	70	SiO ₂	MgO	CaO	Al_2O_3	Fe ₂ O ₃	70
100	82.27	62.43	23.40	44.10	42.04	65.23	17.73	37.57	76.60	55.90	57.96	34.77
150	84.54	66.83	28.04	49.72	44.35	67.89	15.46	33.17	71.96	50.28	55.65	32.11
200	90.20	73.03	38.38	58.54	49.69	73.56	9.80	26.97	61.62	41.46	50.31	26.44
250	98.58	81.59	52.16	72.74	70.31	81.33	1.42	18.41	47.84	27.26	29.69	18.67
300	91.57	62.72	62.93	75.57	71.88	77.2	8.43	37.28	37.07	24.43	28.12	22.8

The component and mass recoveries at different pulp densities given in Figure 4.

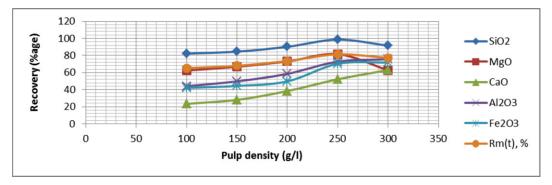


Fig. 4. Component and mass recoveries at different pulp densities.

3.2.3. Optimization of the collector dosage effect in talc flotation test

Five flotation tests were conducted for optimization of collector dosage at different dosage of the mixture of oleic acid and kerosene oil (mixed in equal proportions) used as collector and at constant dosage of depressant (1.0 kg/t), pulp density (200 g/l) and ph value (11) as given in Table 4a. This table revealed that the assay value of SiO₂ and MgO decreases, while the percentages of elemental impurities increases with increasing the dosage of collector from 0.6 to 1.4 kg/t. The component recoveries of SiO₂, MgO, and elemental impurities increases and mass recoveries also increases as dosage of collector increases from 0.6 to 1.2 kg/t as shown in Table 4b. The component recovery of SiO₂ and MgO and mass recovery of concentrate decreased and component recoveries of elemental impurities increased when the dosage of collector increases from 1.2 kg/t. This means that the optimum grade and recovery of talc is obtained at dosage of collector 1.2 kg/t.

3.2.4. Optimization of the pulp density effect in talc flotation test

Five flotation tests were conducted at different pulp densities and at optimum dosage of collector, depressant and pH value. Chemical analysis of tests products results are given in Table 5a, which shows that the assay value of SiO_2 and MgO decreases, while the percentages of elemental impurities increases as pulp density increased from 100 to 300 g/l. The component recovery of SiO_2 , MgO and elemental impurities increases and mass recovery also increases with increases in pulp density from 100 to 200 g/l. The optimum results were obtained at pulp density of 200g/l as shown in Table 5b. This was also confirmed by chemical analysis of tailing products.

4. Conclusion

Talc recovery increases up to pH value of 11 but start decreasing as pH value is increased further from 11. Recovery of talc increases as pulp density increased from 100 g/l to 200g/l. Pulp density above 200 g/l resulted in decrease of recovery. Recovery of talc increases at optimum depressant dosage of 1.0 kg/t by using sodium hexa metaphospahte as depressant, but start decreasing when dosage of depressant increase from 1.0 kg/t. Talc recovery with increase in collector dosage up to 1.2 kg/t, by using oleic acid with kerosene oil as collector. The optimum dosage for collector was found as 1.2 kg/t and above this dosage the recovery of talc decreases.

References

- Ahmed, M. M., Ibrahim, G. A., Hassan, M. M., 2007. Improvement of Egyptian talc quality for industrial uses by flotation process and leaching. International Journal of Mineral Processing, 83(3), 132-145.
- Akdemir, T. A., 2005. Flotation and entrainment behavior of minerals in talccalcite separation. Scandinavian Journal of Metallurgy, 34, 241-244.
- Derco, J., Nemeth, Z. O. L. T. Á. N., 2002. Obtaining of high quality talc from talcose rocks: a case study from the Sinec Kokava deposits (Slovakia). Boletin Paranaense de Geociencias, 50, 119-130.
- Ge Li, Z. L., 2013. Compressive use of dolomite-talc ore to prepare talc, nano-MgO and light weight CaCO₃ using acid leaching method. Applied clay science , 86, 145-152.
- Hojamberdiev, M., Arifov, P., Tadjiev, K., Yunhua, X. U., 2010. Characterization and processing of talc-magnesite from the Zinelbulak deposit. Mining Science and Technology, 20(3), 415-420
- Kursun, H., Ulusoy, U., 2006. Influence of shape characteristics of talc mineral on the column flotation behavior. International Journal of Mineral Processing, 78(4), 262-268.
- Muhammad, W., 2011. Work plan of Mingora emerald mine Swat, KPK, Pakistan. Mingora Swat, KPK, Pakistan.
- Siddiqui, R. A., 1993. Minerals and Rock for Industry. Mingora Swat: Geoloical Survey of Pakistan.
- Wills, B. A. 1992. Mineral Processing Technology. (5th, Ed.) England: Camborn school of mines.